

SAFETY DATA SHEET

Revision Date: 28 September 2016

SECTION I - MATERIAL IDENTIFICATION		
Trade/Material Name:	NATURAL URANIUM OXIDE	
Description:	Metal Oxide	
Other Designations:	NU, U, U-238, U3O8, triuranium octaoxide	
Use of the Substance/Preparation:	Natural uranium oxide is formed when natural uranium is oxidized, NU has the same isotope ratio (²³³ U, ²³⁴ U, ²³⁵ U and ²³⁸ U) found in nature.	
	Natural Uranium Oxide refers to an unspecified mixture of U308 with trace amounts of U03 and U02. Typically in a powder form, it is not considered pyrophoric, combustible or flammable.	
	NOTE: Natural Uranium (NU) is regulated by the U.S. Nuclear Regulatory Commission (NRC), or Agreement State, which should be consulted for specific requirements on all aspects of the production and distribution of this radioactive material.	
Supplier:	Manufacturing Sciences Corporation 804 South Illinois Avenue Oak Ridge, TN 37830 Tel: 865-481-0455 Fax: 865-481-3142	
E-Mail Address:	mscinfo@gmisotope.com	
Emergency Telephone Number:	INFOTRAC EMERGENCY RESPONSE USA & Canada (800) 535-5053 International +1 (352) 323-3500	



SECTION II - HAZARDS

Danger



Natural uranium oxide has not been classified to be a carcinogen by the National Toxicology Program (NTP), International Agency for Research on Cancer (IARC), or the Environmental Protection Agency. The American Conference of Governmental Industrial Hygienists (ACGIH) lists uranium and its compounds as a confirmed human carcinogen. The National Institute for Occupational Safety and Health also list uranium and its compounds as a potential occupational carcinogen because of its alpha-emitting radioactivity if it is inhaled, ingested, or injected.

Warning



NOTE: Natural Uranium Oxide (NU) is regulated by the U. S. Nuclear Regulatory Commission (NRC), or Agreement State, which should be consulted for specific requirements on all aspects of the production, handling and distribution of this radioactive material.



SECTION III – MATERIAL INFORMATION

Ingredient Name:	CAS Number:	Percent:
Uranium Oxide (Natural)	7440-61-1	100

	SECTION IV – FIRST AID		
Eye Contact:	Flush immediately, including under the eyelids, gently but thoroughly with flooding amounts of running water for at least 15 minutes.		
Skin Contact:	Immediately wash with soap and water. Decontaminate body surfaces following radiation standards. Be careful not to abrade skin, in order to avoid systemic uptake.		
Inhalation:	Remove exposed person to fresh air and support breathing as needed.		
Ingestion:	Seek professional medical assistance immediately. Never give anything by mouth to someone who is unconscious or convulsing.		
Safety Stations:	Where powders, fines, dusts, or fumes of the metal oxide are likely to be present, make emergency eye wash stations, safety/quick-drench showers, and washing facilities available in the work area. At a minimum, emergency eyewashes and showers should meet the design and performance requirements of the current ANSI Z358.1 Emergency Eyewash and Shower Standard.		

GET MEDICAL HELP FOR ALL EXPOSURES. Seek prompt medical assistance for further treatment, observation, and support after first aid. Follow established procedures including radiation monitoring programs. **NOTE TO PHYSICIAN**: Following significant ingestion, gastric lavage, with 2% bicarbonate solution, is recommended. A 5% bicarbonate solution has been used by some poison control specialists in radiation treatment. Depending on the solubility of the material, follow-up bioassay (urine) sampling can be used to assess the severity of a potential assimilation.



SECTION V - FIRE

Flash Point (Method):	None reported	Limits: LEL %:	None reported	UEL %:	None reported	
NFPA Fire Hazard Symbol Codes: Flammability: 0 Health: 1 Reactivity: 0 Special: NONE **						
** Values determined by ma	anufacturer, not fou	and in NFPA guidebook	references.			
Extinguishing Media:		Use dry chemical Cla fires. Flood large fires Emergency Response	ss D or carbon dio: with water per De e Guidebook Guide	xide to fight s partment of ⊺ e 162.	small uranium Fransportation	
Unusual Fire or Explosion	n Hazards:	Uranium Oxide is not explosion hazards.	combustible and p	oses no fire	risk or	
Special Fire-Fighting Pro	cedures:	Wear an approved se full-face piece operat mode. Use a Class D	elf-contained breath ed in the pressure fire extinguisher.	iing apparatu demand or p	us (SCBA) with a positive-pressure	

SECTION VI – ACCIDENTAL RELEASE

Spill/Leak Procedures:	Accidental leaks or spills of uranium and its compounds must be planned for well in advance before starting any work procedure. Special radiation procedures are required and professional assistance may be needed. Notify safety or health physics personnel, evacuate all non-essential personnel, and provide adequate ventilation. Clean-up personnel need protection against contact with and inhalation of dust or
Waste Management/Disposal:	Follow all applicable federal, state, and/or local regulations governing the disposal of radioactive waste and contaminated materials.



SECTION VII – HANDLING AND STORAGE

Storage Segregation:	Store uranium oxide in closed containers; prevent access by unauthorized personnel. Depending on quantity stored, containers and area may require special placarding or postings based on RQ and dose rate.
Contamination:	Practice good personal hygiene. Always wash thoroughly after using this material. Avoid transferring it from your hands to your mouth while eating, drinking, or smoking. Do not eat, drink, or smoke in work areas. Do not allow exposure of personnel with open wounds or cuts. Use radiation monitoring equipment, if available, responsive to alpha and beta emissions.
Handling/Storage:	Follow established federal and state regulations for the use and storage of radioactive materials.

SECTION VIII- EXPOSURE CONTROLS/ PPE

Ingredient Name:	CAS Number:	Percent:	Exposure Limits:
Uranium Oxide	7440-61-1	ca 100	* OSHA PEL: 0.05 mg/m³, 8-hr TWA
			(Soluble Compounds)
			OSHA PEL: 0.25 mg/m ^{3 ,} 8-hr TWA
			(Insoluble Compounds)
			* ACGIH: 0.2 mg/m³, TLV-TWA
			0.6 mg/m³, Ceiling
			Toxicity Data: Not Listed
			NRC inhalation DAC: 2E-11 μ Ci/ml (occupational), 6E-14 μ Ci/ml air (effluent), insoluble. Weekly intake limit: 10 mg soluble uranium (10 CFR 20 Appendix B)

*Defined for both soluble and insoluble uranium compounds.



Personal Protective Equipment:	
Goggles:	Wear safety glasses with side shields. In dusty environments, wear chemical safety goggles and a face shield, per OSHA eye- and face-protection regulations.
Respirator:	For emergency operations, entry into unknown atmospheres, or atmospheres immediately dangerous to life or health (IDLH), wear a SCBA with a full-face piece operated in the pressure demand (positive pressure) mode. If significant oxide or powder formation has become airborne or the concentrations exceed OSHA, and/or NRC limits, wear a properly fitted NIOSH-approved air-purifying respirator equipped with HEPA cartridges. Because each type of respirator has an assigned protection factor, respirator selection should be done by an industrial hygienist, health physicist, or other qualified individual.
Other:	Wear impervious gloves, boots, aprons, etc., as appropriate, to prevent prolonged or repeated skin contact.
Workplace Considerations:	
Ventilation:	Provide local ventilation as required to maintain exposure below the USNRC DAC (Derived Air Concentration) and OSHA PELs specified in this section.

SECTION IX – PHYSICAL AND CHEMICAL PROPERTIES

Appearance & Odor: Black to gray crystals or powder and mildly radioactive. No odor found.

Boiling point:	NA	Specific Gravity (H ₂ O=I):	8.30
Vapor Density (Air=1):	NA	Vapor Pressure:	NA
Water Solubility (%):	Insoluble	Melting Point:	>1300°C
Evaporation Rate:	NA	% Volatile By Volume:	NA



SECTION X - STABILITY AND REACTIVITY

Chemical Uranium oxide can react with bromine trifluoride rapidly below 275°F. **Incompatibilities:**

Conditions to Avoid: Do not create dusty work conditions.

Hazardous Decomposition Products:	Radioactive progeny (daughters) thorium-234, protactinium-234, and -234m (metastable) are produced by natural radioactive decay; they are the source of the majority of the penetrating radiation.
Stability/Polymerization:	Material is stable in closed containers at room temperature under normal storage and handling conditions. Hazardous polymerization cannot occur.

SECTION XI – TOXICOLOGICAL

Summary of Risk:	Uranium oxide is both toxic and radioactive. Dermatitis, renal damage, acute necrotic arterial lesions, and possibly death may occur from extreme exposure. Inhalation of fine uranium particles presents increased radiation hazards; isolated uranium particles in the lungs may be a long-term cancer hazard. The more soluble uranium compounds are considered most toxic to the kidneys; the lung is the critical organ for insoluble respirable dusts or fines such as oxide powders. Uranium dusts are respiratory irritants, with coughing, shortness of breath as possible outcomes. Prolonged skin contact can cause damage to the basal cells. Radioactivity is the property of the spontaneous emissions of alpha or beta particles and gamma rays, by the disintegration of the nuclei of the atoms.
Medical Conditions Which May Be Aggravated by Contact:	None reported.
Target Organs:	Respiratory system; skin; eyes; kidneys; liver; blood; lymphatic system; and bone marrow.



Primary Entry Route(s):

Acute Effects:

Chronic Effect(s):

In solid forms, ingestion, skin, or eye contact. Inhalation of dusts or fines.

Nausea, vomiting, shortness of breath, and coughing.

Primarily the effects of radiation from insoluble compounds. Possibilities include pneumoconiosis, pulmonary fibrosis, lymphoma, osteosarcoma, and lung cancer.

GET MEDICAL HELP (INPLANT OR COMMUNITY) FOR ALL EXPOSURES. Seek prompt medical assistance for further treatment, observation, and support after first aid. Follow established procedures including radiation monitoring programs. **NOTE TO PHYSICIAN:** Following significant ingestion, gastric lavage, with 2% bicarbonate solution, is recommended. A 5% bicarbonate solution has been used by some poison control specialists in radiation treatment. Depending on the solubility of the material, follow-up bioassay (urine) sampling can be used to assess the severity of a potential assimilation.

Carcinogenicity:

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SECTION XII – ECOLOGICAL

Uranium oxide is present naturally in virtually all soil, rock and water. Uranium in soil and rocks is distributed throughout the environment by wind, rain and geologic processes. Rocks weather and break down to form soil, and soil can be washed by water and blown by wind, moving uranium into streams and lakes, and ultimately settling out and reforming as rock. Uranium can also be removed and concentrated by people through mining and refining. These mining and refining processes produce wastes such as mill tailings which may be introduced back into the environment by wind and water if they are not properly controlled. Manufacturing of nuclear fuel, and other human activities also release uranium to the environment.

Information provided by United States Environmental Protection Agency

SECTION XIII – DISPOSAL

Follow all applicable federal, state, and/or local regulations governing the disposal of radioactive waste and contaminated materials.



SECTION XIV – TRANSPORTATION

Transportation Data (49 CFR 172.101-2):

Proper Shipping Name:

Identification Number:

DOT Hazard Class:

Radioactive Material, Low Specific Activity, n.o.s.

UN 2912

Radioactive Material, Class 7 UN Register: UN 2912

SECTION XV – REGULATORY

OSHA Designations:

Listed as Air Contaminant (29 CFR 1910.1000)

EPA Designations:

RCRA Hazardous Waste (40 CFR 261.33): Exempted from RCRA regulation as Source Material.

SARA Extremely Hazardous Substance (40 CFR 355): Not listed.

SARA Section 313 Toxic Chemical (40 CFR 372.65): Not listed.

Reported in EPA TSCA Inventory: Yes

U.S. Nuclear Regulatory Commission Designations:

Natural Uranium (NU) is regulated by the NRC, or Agreement State, which should be consulted for specific requirements on all aspects of the production, handling and distribution of this radioactive material.

SECTION XVI – OTHER INFORMATION

Prepared By:

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